## A strategy for the stepwise ring annulation of all four pyrrolic rings of a porphyrin<sup>†</sup>

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The repeated introduction of an  $\alpha$ -dione unit and its reaction with an arene-1,2-diamine allows the stepwise annulation of all four pyrrolic rings of a porphyrin, as is demonstrated by the synthesis of a trisquinoxalinoporphyrin, a tetrakisquinoxalinoporphyrin and the more elaborated bisporphyrin.

Porphyrins with  $\beta$ ,  $\beta'$ -pyrrolic bonds that are fused with aromatic rings have applications in a variety of fields from molecular electronics to biomedicine.<sup>1-3</sup> The syntheses of the simplest tetraannulated representatives, the tetrabenzoporphyrins<sup>4,5</sup> and tetranaphthaloporphyrins,<sup>6</sup> have been reported. The methodology for their construction, however, is not amenable for making more elaborate arrays or for stepwise building-up of square grid networks of porphyrins (Fig. 1), which might find use in molecular circuitry and other molecular electronics applications. We now outline a strategy that allows the stepwise ring annulation of each of the four pyrrolic rings of a porphyrin. The key to this approach is the successive introduction into each ring of a functionality that can be elaborated readily. We have chosen to expand an approach to ring annulation outlined in the synthesis of much simpler extended porphyrin systems, in which the ring fusion was created by the reaction of a porphyrin- $\alpha$ -dione with an arene-1,2 -diamine.7

Our approach is illustrated by the synthesis of end-capped trisquinoxalinoporphyrin 1 and tetrakisquinoxalinoporphyrin 2,



Fig. 1 Stepwise functionalisation of porphyrin pyrrolic rings allows controlled mono-, *syn*-bis-, *anti*-bis-, tris- and tetrakis-annulation.

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which are extended in three and four directions, respectively, and the bis-porphyrin 3, in which the possibilities for constructing extended porphyrin arrays are demonstrated. Systems 1 and 2, which are laterally-extended porphyrins with an enlarged  $\pi$ -electronic framework, are interesting compounds in their own right.



In previous work, we have developed the syntheses of porphyrin-2,3-diones from 2-aminoporphyrins and 2-hydroxyporphyrins, and porphyrin-2,3,7,8-tetraones and porphyrin-2,3,12,13-tetraones from diaminoporphyrins.<sup>7,8</sup> These have been used to create extended porphyrin systems for a variety of uses.<sup>7–11</sup> Our synthesis of the tetraones<sup>7</sup> involved initial dinitration of the porphyrin periphery, which led to a mixture of isomers that were carried forward to common products. Porphyrins with two functionalised pyrrolic rings are also accessible in a stepwise manner by the nitration of porphyrin-2,3-diones and further elaboration, and by similar sequences on quinoxalino[2,3-*b*]porphyrins.<sup>12</sup> The latter compounds serve as models of more elaborate

systems and allow the introduction of two different annulated ring systems, for example one appending a  $C_{60}$  and the other a ferrocene.<sup>11</sup> The route to compounds 1 and 2 is outlined in Scheme 1. Corner zinc(II) porphyrin-tetraone<sup>7</sup> 4 was functionalised further at one or both  $\beta$ , $\beta'$ -pyrrolic faces for the synthesis of trisquinoxalinoporphyrin 1 and tetrakisquinoxalinoporphyrin 2.‡

Trisquinoxalinoporphyrin 1 was prepared by first reacting 4 with *ortho*-phenylenediamine in  $CH_2Cl_2$  solution to afford the corner zinc(II) bisquinoxalinoporphyrin 5 in 90% yield. Nitration of 5 afforded an inseparable mixture of corner zinc(II) 12- and 13nitro-bisquinoxalinoporphyrins 6 in 83% yield, but both gave the same dione subsequently. Demetallation of 6 was achieved using HCl, affording corner 12- and 13-nitro-bisquinoxalinoporphyrins 7 in 89% yield. The tin(II)-mediated reduction of 7 gave the corresponding corner 12- and 13-amino-bisquinoxalinoporphyrins 8, which were photo-oxidised in the presence of *ortho*-phenylene-diamine for 5 h to give trisquinoxalinoporphyrin 1 in 64% yield. The intermediate corner bisquinoxalinoporphyrin-12,13-dione 9 was formed as an intermediate. The overall yield from 4 of 1 was 47%.

Corner zinc(II) porphyrin-tetraone **4** also served as a precursor to tetrakisquinoxalinoporphyrin **2**. Nitration of **4** in gave an isomeric mixture of corner zinc(II) 12,17-, 12,18- and 13,17-dinitroporphyrin-tetraones **10** in 59% yield. Reaction of **10** with *ortho*-phenylenediamine gave the corner zinc(II) 12,17-, 12,18- and 13,17-dinitro-bisquinoxalinoporphyrins **11** in 66% yield.



Scheme 1 Reagents and conditions: (i) ortho-phenylenediamine in  $CH_2Cl_2$ , 5 min; (ii)  $NO_2$  in light petroleum,  $CH_2Cl_2$ ; (iii)  $HCl-CH_2Cl_2$ ; (iv)  $SnCl_2\cdot 2H_2O$  in  $HCl-Et_2O$ ; (v)  $O_2$  and ortho-phenylenediamine in  $CH_2Cl_2$ , hv, 5 h; (vi)  $O_2$  and ortho-phenylenediamine in  $CH_2Cl_2$ , hv, 45 min; (vii)  $Zn(OAc)_2\cdot 2H_2O$  in  $CHCl_3-CH_3OH$ , heat, 5 min; (viii)  $O_2$  and ortho-phenylenediamine in  $CH_2Cl_2$ , hv, 20 min.



Scheme 2 Reagents and conditions: (i)  $O_2$  in CH<sub>2</sub>Cl<sub>2</sub>, hv, 80 min; (ii) HCl–CH<sub>2</sub>Cl<sub>2</sub>, 2 h, followed by SnCl<sub>2</sub>·2H<sub>2</sub>O in HCl–Et<sub>2</sub>O, 10 min; (iii) toluene, reflux, 36 h.

Treatment of 11 with a HCl-CH<sub>2</sub>Cl<sub>2</sub> solution afforded free base corner 12,17-, 12,18- and 13,17-dinitro-bisquinoxalinoporphyrins 12 in 83% yield. The isomeric mixture 12 was reduced using SnCl<sub>2</sub>·2H<sub>2</sub>O in HCl-Et<sub>2</sub>O to afford the diamino analogues 13, which were then photo-oxidised in the presence of orthophenylenediamine to afford amino-trisquinoxalinoporphyrin 15 in 83% yield by way of dione 14. The amino-trisquinoxalinoporphyrin 15 was converted into its zinc(II) derivative 16. Photooxidation of 16 gave zinc(II) trisquinoxalinoporphyrin-dione 17. which was reacted in situ with ortho-phenylenediamine to afford tetrakisquinoxalinoporphyrin **18** in 84% zinc(II) vield. Demetallation of 18 by treatment with HCl in CH<sub>2</sub>Cl<sub>2</sub> gave tetrakisquinoxalinoporphyrin 2 in 93% yield. The overall yield of 2 from 4 was 22%.

The complete demetallation of zinc(II) tetrakisquinoxalinoporphyrin **18** by HCl treatment required 5 h, in contrast to simple zinc(II) porphyrins, which are demetalled in a matter of seconds with this reagent. This difference is an indication that the quinoxalino substituents are able to exert a strong electronwithdrawing effect on the porphyrin ring, especially under acidic conditions.

In order to illustrate how the synthetic strategy outlined here can be utilised for the preparation of more elaborated systems, bis-porphyrin **3**, in which all pyrrolic rings are annulated, was synthesised. The photo-oxidation of **17** in the presence of 1,2,4-triamino-5-nitro-benzene afforded zinc(II) amino-nitrotetrakisquinoxalinoporphyrin **19** in 65% yield (Scheme 2). The conversion of **19** into the diamino-tetrakisquinoxalinoporphyrin **20** in 78% yield was achieved by acidic demetallation followed by reduction using  $SnCl_2 \cdot 2H_2O$ . The condensation of diamine **20** and trisquinoxalinoporphyrin-dione **21**, obtained by the



Fig. 2 Electronic absorption spectra in  $CHCl_3$  of trisquinoxalinoporphyrin 1 (blue), tetrakisquinoxalinoporphyrin 2 (red) and bis-porphyrin 3 (green).

photo-oxidation of 15, afforded the elaborated bis-porphyrin 3 in 62% yield.

Increasing the number of quinoxaline units fused to the porphyrin macrocycle results in a widening of the light absorption window and a red-shift in electronic spectra compared to simple porphyrins due to the increase in the  $\pi$ -electronic network. For example, **2** and **3** exhibit a wide absorption window, with Q band maxima up to 697 and 735 nm, respectively (Fig. 2). Such broad absorption properties might find use in artificial light harvesting systems and light detectors.

In this work, a strategy for the step-wise annulation of all four pyrrolic rings of a porphyrin has been demonstrated, and a synthetic protocol has been established that will enable the generation of more complex systems that may be of value in applications from molecular electronics to artificial light harvesting systems.

## Notes and references

‡ All new compounds have been fully characterised, including elemental analysis and/or high resolution ESI-FT/ICR.

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